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Roll No.:

337353(37)

B. E. (Third Semester) Examination, Nov.-Dec. 2021

(New Scheme)

(Mech. Engg. Branch)

MATERIAL SCIENCE & METALLURGY

Time Allowed: Three hours

Maximum Marks: 80

Minimum Pass Marks: 28

Note: Attempt all questions. Part (a) of each question is compulsory. Attempt any two parts from the rest of three parts (b), (c) and (d) in each question.

Unit-I

- 1. (a) Define atomic packing factor.
- _
- (b) Discuss effect of grain size on properties of metals. 7

	[2]				[3]	
(c)	Discuss mechanism of solidification of metals.	7			find the percentage amount of solid and liquid phase	
(d	Explain Homogeneous and Heterogeneous nucleation				portion of the diagram by Lever Rule.	7
	in short.	7		(d)	Draw and explain Iron-Carbon equilibrium diagram.	7
	and one and mount-II programs bould to I				Unit-IV	
2. (a)	What is Elastic Deformation?	2	4.	(a)	Define "Annealing".	2
(b) Define Stiffness, Yield strength, Resilence and				Explain T-T-T curve in brief.	7
	Ductility.	7		(c)	What are the purposes of Heat treatment & explain	
(c)	Compare Edge and Screw dislocation.	7			Normalizing?	7
(d)	Define Interstitial defect, Franket defect & Schottky			(d)	Explain the process of Carbo-Nitriding and Sphero-	
	defect.	7			idising process.	7
	Unit-III				Unit-V	
3. (a)	What is Gibb's phase rule?	2	5.	(a)	Enumerate the various ores which yield ferrous	
(b)	Explain the allotropy of iron alongwith a neat sketch				metals.	2
	indicating the different phases for pure iron at various temperatures.	7		(b)	Give the composition properties and uses of any of the two :	7
(c)	Draw equilibrium diagram when two metals				(i) White Cast Iron	
	completely soluble in the liquid and solid state and				(ii) High Carbon Steel (iii) Nodular Cast Iron	

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(C)	Give the composition, properties and uses of any of	
	the two:	7
	(i) Naval Brass	
	(ii) Gun metal	
	(iii) Sintered Carbide	
<i>(</i> 1)	TATE A 1 A 1 A 1 A 1 A 1 A 1 A 1 A 1 A 1 A	
(a)	What is the contribution of light metals and their alloys for solving the problems related to corrosion?	
	Give examples.	7
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